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# Synthesis and pharmacological activity of 1,3,6-trisubstituted-4-oxo-1,4-dihydroquinoline-2-carboxylic acids as selective $ET_A$ antagonists

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#### ABSTRACT

A series of 1,3,6-trisubsituted-4-oxo-1,4-dihyroquinoline-2-carboxylic acid analogs (2a-m) were designed and synthesized and their pharmacological activity determined, with the objective to better understand their SAR as potential  $ET_A$  selective inhibitors. Most of the compounds displayed significant  $ET_A$  antagonist activity having  $IC_{50}$  for inhibition of binding of the [ $^{125}I$ ]ET-1 to  $ET_A$  receptor <10 nM, with good selectivity for  $ET_A$  antagonism over  $ET_B$  receptor. Based on the in vitro results, SAR of this series of compounds requires an alkoxy substituent at the 6-position to be a straight and saturated chain up to three carbons long, since substitution of unsaturated and branched alkyloxy groups results in decrease in  $ET_A$  antagonist activity. In this series, compound  $ET_A$  (6-0-n-propyl analog) was found to be most potent ( $ET_A$ ) = 0.11 nM) with  $ET_B/ET_A$  selectivity of 8303.

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Endothelins are divided into three structurally similar subtypes: endothelin-1, 2, and 3 (ET-1, ET-2, and ET-3), which are members of 21 amino acid peptide families. Among them, ET-1 is the major isoform involved in human cardiovascular and non-cardiovascular physiology and pathophysiology. ET-1 is a very potent vasoconstrictor, 10 times more potent than angiotensin II and 100 times more potent than epinephrine and also has inotropic, pro-inflammatory, and mitogenic properties. It also influences the salt-water homeostasis, and stimulates the renin–angiotensin–aldosterone and sympathetic nervous systems. Increased level of the plasma concentration of ET-1 has been implicated in several diseases such as hypertension, pulmonary hypertension, acute myocardial infarction, congestive heart failure, renal failure, atherosclerosis, pulmonary fibrosis, and cancer.

ET-1 exerts its physiological action by binding to the specific receptors called endothelin receptor type A (ET<sub>A</sub>) and B (ET<sub>B</sub>), belonging to the heptahelical G-protein-coupled receptor superfamily.<sup>5</sup> Endothelin receptors are widely expressed in most of the physiological systems, and interaction of ET<sub>A</sub> expressed on VSMC with ET-1 results in vasoconstriction and cell proliferation.<sup>6</sup> On the other hand, ET<sub>B</sub> activation is mainly responsible for vasodilation depending on tissue location<sup>7</sup> and clearance of endogenous endothelins.<sup>8</sup>

Endothelin receptor antagonists (ERAs), both selective  $ET_A$  antagonist and non-selective  $ET_A/ET_B$  antagonists, are novel thera-

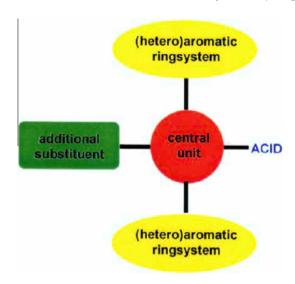
peutics in clinical development for various diseases like pulmonary arterial hypertension, chronic heart failure, vascular remodeling (restenosis, atherosclerosis), renal failure, cancer, and lung fibrosis. Though theoretically it seems, selective ET<sub>A</sub> receptor antagonists would be useful as therapeutic agents for the treatment of the aforementioned chronic pathological conditions. Advantages of a selective antagonist of the ET<sub>A</sub> receptor <sup>10</sup> are:

- Vasoconstriction in the human cardiovascular system by ET-1 appears to be predominantly mediated by the ET<sub>A</sub> receptor.
- The mitogenic and pro-inflammatory effects induced by ET-1 are mediated by ET<sub>A</sub> receptor stimulation.
- Blockade of the ET<sub>B</sub> receptor would prevent the putative beneficial effects like vasodilation of ET-1-induced NO and PGI<sub>2</sub> release from the endothelium.
- Since the ET<sub>B</sub> receptor in the lung is involved in clearance of ET-1, ET<sub>B</sub>-antagonism will reduce the rate of clearance of exogenous ET-1 from the circulation.

Most non-peptidic endothelin receptor antagonists have the following common structural features: a carboxylic acid group or an acidic sulfonamide moiety between two aromatic or hetero-aromatic ring systems (Fig. 1).<sup>10</sup> These pharmacophore groups are positioned so that they mimic amino acids of the ET-1 C-terminal, which is in a helical form. Different central core units like indane (enrasentan),<sup>11</sup> chromene,<sup>12</sup> cyclopentenopyridine (L-753037),<sup>13,14</sup> and pyrrolidine (atrasentan)<sup>15</sup> have been incorporated into endothelin antagonists currently in the clinical trials (Fig. 2).

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**Figure 1.** General structure of endothelin receptor antagonists. <sup>10</sup>

Figure 2. Examples of endothelin receptor antagonists in clinical trials.

Earlier, when the indane ring of SB-209670 was superimposed on the quinolone ring, a good overlap was obtained with 1,3-disubstituted-2-carboxylic quinolones (1). Preliminary SAR resulted in compound 1 having potent ET<sub>A</sub> antagonist activity and was 800 times more selective for ET<sub>A</sub>. <sup>16</sup> Also, it had been reported that introduction of an alkylamino or alkoxy was beneficial to potency and selectivity in analogs belonging to indane, <sup>11</sup> chromene, <sup>12</sup> and cyclopenteno[1,2*b*]pyridine<sup>13,14</sup> carboxylic acid series. Therefore it was hypothesized that introducing an alkoxy substituent at the 6th position of the A ring of quinolone would lead to a new series of compounds with increased potency and selectivity towards ET<sub>A</sub> receptor (Fig. 3). This hypothesis was validated by generating the CoMFA model of endothelin receptor antagonists having IC<sub>50</sub> values between 0.19 and 780 nM, which showed steric groups favorable at the 6 position of the quinolone ring.

In the synthesis of the target compounds 1,3,6-trisubstituted-4-oxo-1,4-dihydro-quinoline-2-carboxylic acids (**2a–m**), compound **15** is the key intermediate. Alkylation of **15** with various alkyl halides led to analogs with different substituents at the 6-position of

**Figure 3.** Hypothesis for the synthesis of 1,3,6-trisubstituted-4-oxo-1,4-dihydro-quinoline-2-carboxylic acid as selective ET<sub>A</sub> antagonists.

the quinolone. As shown in Scheme 1, nitration of 3-hydroxyacetophenone with nitric acid in the presence of acetic acid resulted in the desired regioisomer 3-hydroxy-6-nitroacetophenone, **4**, which had the reported melting point. Compound **4** was condensed with 3-carboxybenzaldehyde to yield *E*-configuration chalcone **5**. Carboxylic acid **5** was converted into ester **6** by reaction with anhydrous ethanol saturated with HCl gas followed by MOM protection of the hydroxyl group resulting in compound **7**. Hydrogenation at 45 psi with PtO<sub>2</sub> catalyst for 45 minutes reduced both the nitro group and the double bond of chalcone **7** to give amine **8**.

For the synthesis of intermediate 6-ethylpiperonal, **11**, Scheme 2 was followed. Wolff–Kishner reduction of 3,4-methylenedioxyace-tophenone (**9**) resulted in crude product, which was purified by vacuum–distillation at 130 °C and 27 mm/Hg to yield 63.4% of pure **10**. Formylation of 3,4-methylenedioxy ethylbenzene (**10**) with  $\alpha,\alpha$ -dichloromethyl methyl ether as precursor of the formyl group, via alkylation catalyzed by titanium tetrachloride, resulted in 6-ethylpiperonal (**11**) in excellent yield. <sup>19</sup>

Reductive amination of 6-ethylpiperonal, **11** with aromatic amine **8** was then carried out by using hydride reducing agents such as sodium cyanoborohydride (NaBH<sub>3</sub>CN),<sup>20</sup> or titanium (IV) isopropoxide (Ti(OiPr)<sub>4</sub>)/sodium cyanoborohydride (NaBH<sub>3</sub>CN),<sup>21</sup> or sodium triacetoxyborohydride (NaBH(OAc)<sub>3</sub>).<sup>22</sup> Among these three methods, reductive amination with NaBH(OAc)<sub>3</sub> gave the best yield of product **12**. N-acylation of **12** with ethyl oxalyl chloride gave an excellent yield of compound **13**, which on cyclization by the hindered base 1,8-diazabicyclo-[5.4.0]undec-7-ene (DBU) in ethanol gave MOM-protected-4-quinolone diester **14**. Deprotection of MOM-group under acidic condition resulted in key intermediate **15** (Scheme 3).

Alkylation of **15** with various alkyl halides in presence of sodium hydride and potassium carbonate led to derivatives with different substituents at the 6-position of the quinolone **16a-k** (Scheme 4). The synthesis of the target compounds (**2a-m**) was accomplished by potassium hydroxide treatment of the 4-oxoquinoline diesters in moderate yields. The general synthetic scheme along with the physical properties is described in Supplementary data.

All compounds listed in Table 1 were evaluated for their ability to inhibit endothelin-A (ET<sub>A</sub>) and endothelin-B (ET<sub>B</sub>) receptors in vitro. The results are summarized in Table 1. For each compound, eight concentrations were tested in the presence of fixed concentration of radio-labeled ET-1([ $^{125}$ I] ET-1) on green monkey renal tubular (Vero) cells expressing ET<sub>A</sub> receptors  $^{23}$  and membrane expressing ET<sub>B</sub> receptor. Besides thirteen proposed compounds, BQ-123 (selective ET<sub>A</sub> antagonist) was taken as a positive control to validate the method. The IC50 values, which are the concentrations that reduce the [ $^{125}$ I] ET-1 binding to the

Scheme 1. Preparation of intermediate 8. Reagents and conditions: (a) HNO<sub>3</sub>, glacial CH<sub>3</sub>COOH, 70 ± 2 °C; (b) 3-carboxybenzaldehyde, 10 N NaOH (3 equiv), CH<sub>3</sub>OH, reflux; (c) dry HCl gas, anhydrous C<sub>2</sub>H<sub>5</sub>OH; (d) MOMCl, NaH, dry K<sub>2</sub>CO<sub>3</sub>, dry DMF; (e) 45 psi H<sub>2</sub> gas, PtO<sub>2</sub>, anhydrous CH<sub>3</sub>OH.

**Scheme 2.** Synthesis of 6-ethylpiperonal, **11**. Reagents and conditions (a) (i)  $NH_2NH_2\cdot H_2O$ , ethanol, reflux, (ii) KOH, 160 °C; (b) TiCl<sub>4</sub>,  $\alpha,\alpha$ -dichloromethyl methyl ether, dichloroethane, 0 °C to rt.

Scheme 3. Preparation of intermediate 15. Reagents: (a) NaBH(OAc)<sub>3</sub>, DCE; (b) ethyl oxalyl chloride, NEt<sub>3</sub>, dry DMF; (c) DBU, ethanol; (d) dry HCl gas, anhydrous C<sub>2</sub>H<sub>5</sub>OH.

receptors by 50%, were determined for each drug and are listed in Table 1.

On the basis of this study, the following structure–activity relationships are proposed for the 1,3,6-trisubstituted-4-oxo-1,4-dihydro-quinoline-2-carboxylic acid analogs,  ${\bf 2a-m}$  as an ET<sub>A</sub> antagonists. Among the compounds tested, 6-O-ethyl ( ${\bf 2b}$ ), 6-O-n-propyl ( ${\bf 2c}$ ), 6-O-n-butyl ( ${\bf 2g}$ ), 6-O-i-butyl ( ${\bf 2h}$ ), 6-O-sec-butyl ( ${\bf 2i}$ ), 6-O-propan-3-ol ( ${\bf 2l}$ ), 4-oxo-1,4-dihydroquinoline diacids displayed significant ET<sub>A</sub> antagonist activity having IC<sub>50</sub> for inhibition of binding of the [ $^{125}$ I]ET-1 to ET<sub>A</sub> receptor <10 nM. As shown in Table 1, when the 6-OH group of  ${\bf 2a}$  is replaced by a 6-alkoxy group of

two carbons (**2b** IC<sub>50</sub>: 6.6 nM), there is a sixfold increase in ET<sub>A</sub> antagonist activity and a slight increase in selectivity for ET<sub>A</sub> over ET<sub>B</sub>, i.e., 473 for **2b** versus 399 for **2a**. Increase in chain length by one more carbon to 3 carbons (**2c**) resulted in 60-fold increase in ET<sub>A</sub> (IC<sub>50</sub>: 0.11 nM) and threefold increase in ET<sub>B</sub> (IC<sub>50</sub>: 913 nM) antagonist activity compared to **2b**. Although there is increase in binding affinity for both the receptors, a n-propoxy group at 6-position of quinoline is more beneficial for ET<sub>A</sub> receptor binding as seen in selectivity of more than 8000 for ET<sub>A</sub> over ET<sub>B</sub>. There is 10-fold decrease in ET<sub>A</sub> antagonist activity compared to **2c** when the side chain length is increased by one carbon to butyl (**2g**;

$$\begin{array}{c} \text{COOC}_2\text{H}_5 \\ \text{N} \\ \text{COOC}_2\text{H}_$$

**Scheme 4.** Synthesis of 6-O-substituted-4-oxo-quinoline-dicarboxylic acids (**2a-m**). Reagents and conditions: (a) alkyl halides, NaH, dry K<sub>2</sub>CO<sub>3</sub>, dry DMF; (b) 6 N KOH (6 equiv), C<sub>2</sub>H<sub>5</sub>OH, reflux.

Table 1 In vitro inhibitory binding assay ( $IC_{50}$ ) of 1,3,6-trisubstituted-4-oxo-1,4-dihydroquinoline-2-carboxylic acids, **2a-m** 

No.	R	Mp (°C)	$ET_A IC_{50}^a (nM)$	$ET_B IC_{50}^b (nM)$	ET <sub>B</sub> /ET <sub>A</sub>
2a	Н	198-199	38.9 ± 2.6	15561.5 ± 162.7	399
2b	$C_2H_5$	213-215	6.6 ± 1.1	3119 ± 54.3	473
2c	n-C <sub>3</sub> H <sub>7</sub>	162-163	$0.11 \pm 0.02$	$913.4 \pm 27.2$	8304
2d	i-C <sub>3</sub> H <sub>7</sub>	199-200	13.5 ± 1.5	677 ± 9.6	50
2e	CH <sub>2</sub> =CH-CH <sub>2</sub>	165-166	38.1 ± 3.6	1131 ± 24.8	30
2f	$CH_3-O-CH_2$	176–177	24.4 ± 2.9	$1030 \pm 17.3$	43
2g	n-C <sub>4</sub> H <sub>9</sub>	161-162	$1.44 \pm 0.6$	548.2 ± 14.5	381
2h	i-C <sub>4</sub> H <sub>9</sub>	229-231	6.3 ± 1.5	1983 ± 35.7	315
2i	sec-C <sub>4</sub> H <sub>9</sub>	185-186	9.9 ± 1.7	_	_
2j	cyclo-C <sub>3</sub> H <sub>5</sub> -CH <sub>2</sub>	173-174	115.9 ± 11.4	_	_
2k	n-C <sub>5</sub> H <sub>11</sub>	153-155	610 ± 54	_	_
21	HO-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	192-194	$7.4 \pm 0.9$	4387 ± 63.7	593
2m	HOOC-CH <sub>2</sub>	177-180	284.5 ± 3.5	1453 ± 20.7	5
	BQ123	_	$24.8 \pm 1.3^{\circ} (25)$	_	_

<sup>&</sup>lt;sup>a</sup> Monkey renal cells CCL-81 grown as monolayer.<sup>23</sup> n = 3.

IC<sub>50</sub>: 1.6 nM), but still shows better activity than **2b**. Further increase in length to 5 carbons (**2k**; IC<sub>50</sub> 610 nM) resulted in 6000-fold loss in activity compared to 6-O-n-propyl analog (**2c**), also seen in the cyclopenteno[1,2-b]pyridine series as ET<sub>A</sub> receptor antagonist.<sup>14</sup>

Replacement of the saturated 6-O-n-propyl group with the unsaturated 6-O-allyl group resulted in  $\mathbf{2e}$  analog ( $\mathbf{1C}_{50} = 38.11$  nM) 380 times less active than the n-propyl analog  $\mathbf{2c}$ . This suggests that a saturated alkyl chain is required at the 6-position of the 4-oxo-1,4-dihydroquinoline diacids for better  $\mathbf{ET}_A$  antagonist activity. Also, when the second atom of the n-propyl chain in  $\mathbf{2c}$  is replaced by isosteric O ( $\mathbf{2f}$ ;  $\mathbf{1C}_{50}$ : 24.4 nM), there is 240-fold decrease in activity compared to  $\mathbf{2c}$ .

Substitutions at the C1 of the 6-O-alkyl chain are not well tolerated. There is a twofold decrease in  $ET_A$  antagonist activity

compared to **2b** when the methylene group of the ethyl chain is substituted, **2d** (*i*-propyl,  $IC_{50} = 13.5 \text{ nM}$ ). Similarly, substitution of the methylene group at C1 of the 6-*O*-*n*-propyl chain resulted in compound **2i** (6-*O*-sec-butyl) which is 90-fold less active, than **2c**, in inhibition of binding of the [ $^{125}I$ ]ET-1 to ET<sub>A</sub> receptor ( $IC_{50} = 9.9 \text{ nM}$ ). Likewise, branching at C2 of the alkyl chain is not tolerated, as substitution of a methyl group at C2 of the 6-*O*-*n*-propyl chain resulted in analog **2h** (6-*O*-i-butyl) which is 60 times less active as ET<sub>A</sub> receptor antagonist ( $IC_{50} = 6.6 \text{ nM}$ ) than **2c**. Also, compound **2j** (6-*O*-cyclopropylmethyl) which is a cyclic analog of **2h**, shows 1000 times less ET<sub>A</sub> receptor antagonist activity than **2c**. These results suggest that straight chain saturated alkoxy groups are required at the 6-position of the 4-oxo-1,4-dihydroquinoline diacids for good ET<sub>A</sub> receptor antagonism.

b Endothelin-B receptor precursor (human) membranes. n = 2.

<sup>&</sup>lt;sup>c</sup> Literature IC<sub>50</sub> value, *ibid*.

Introduction of a hydroxyl group (2I; IC<sub>50</sub>: 7.4 nM) at C3 of the 6-*O*-*n*-propyl side chain resulted in 70-fold reduction in ET<sub>A</sub> receptor antagonist activity compared with 2c. Also, substitution of a carboxylic acid group (2m; IC<sub>50</sub>: 284.46 nM) for a methyl group in 2b causes a 43-fold decrease in binding affinity for ET<sub>A</sub> receptor compared with 2b. These results are consistent with the previously reported SAR of cyclopenteno[1,2-*b*]pyridine analogs as ET<sub>A</sub> receptor antagonists.<sup>14</sup>

Compound  ${\bf 2a}$  was found to reduce lung inflammation induced by lipopolysaccharide (LPS),  $^{24}$  cigarette smoke, and ET-1 $^{25}$  in a hamster animal model by reducing the recruitment of broncoalveolar lavage fluid (BALF) leukocytes (neutrophils and macrophages). Therefore, reducing ET-1 activity with selective ET<sub>A</sub> receptor antagonists could potentially decrease neutrophil derived enzymes and oxidants in the lung, thereby decreasing or inhibiting the progression of lung injury. Compound  ${\bf 2a}$  was also effective in controlling preterm delivery induced by lipopolysaccharide (LPS) in a pregnant mouse animal model.  $^{26}$  This result, in accordance with previous literature, suggests the role of ET-1 in both normal parturition and premature birth.

In summary, a new series of 1,3,6-trisubstituted-4-oxo-1,4dihydroquinoline-2-carboxylic acid analogs were synthesized and evaluated for their ET<sub>A</sub> antagonist activity and selectivity for ET<sub>A</sub> receptor over ET<sub>B</sub> receptor. Among the compounds tested, **2b**, **2c**, 2g, 2h, 2i, and 2l 4-oxo-1,4-dihydroquinoline diacids displayed significant ETA antagonist activity with IC50 values of inhibition of binding of the [125] ET-1 to ET<sub>A</sub> receptor <10 nM. These compounds also showed good selectivity for ET<sub>A</sub> antagonism over ET<sub>B</sub> receptor. Initial SAR at C6 of the quinolone ring indicates straight, unbranched, and saturated alkyl chain up to three carbons long, with hetero atom spacer required for good ETA antagonist activity and selectivity. Also, polar hydrophilic groups at the 6-position are unfavorable for ETA antagonist activity, as seen with a decrease in activity when substituted by 6-O-propan-3-ol (21) and 6-O-carboxymethyl (2m) of the quinoline ring. The 6-0-n-propyl-4-oxo-1,4-dihydroquinoline diacids analog, 2c was found to be most potent (IC<sub>50</sub>: 0.11 nM) with ET<sub>B</sub>/ET<sub>A</sub> selectivity of 8303. In vivo, compound **2a** (6-OH), a prototype of this 4-oxo-1.4-dihydroguinolone diacid series was shown to reduce LPS-induced lung inflammation and preterm labor in two different animal models.

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### Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/i.bmcl.2010.08.074.

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